



Stereospecific [1,2]-rearrangement of a spirocyclic ammonium ylide with ring expansion sequence

Antonio Saba*

Dipartimento di Chimica, Facoltà di Scienze, Via Vienna 2, I-07100 Sassari, Italy

Received 26 November 2002; revised 13 February 2003; accepted 14 February 2003

Abstract—Enantiomerically pure bicyclic 1,4-oxazepinone was obtained by the Cu(II)-catalyzed decomposition of an α -diazocarbonyl compound tethered to a chiral morpholinone, through the cascade evolution of the spirocyclic ammonium ylide formed. LiAlH_4 reduction and transesterification of the lactone moiety of the oxazepinone afforded pure chiral pyrrolidine and 3-prolinone bicyclic hemiacetal, respectively, both bearing a quaternary stereocentre. © 2003 Elsevier Science Ltd. All rights reserved.

The tandem ammonium ylide generation/rearrangement sequence has a great potential as a simple method for a rapid preparation of nitrogen heterocycles.¹

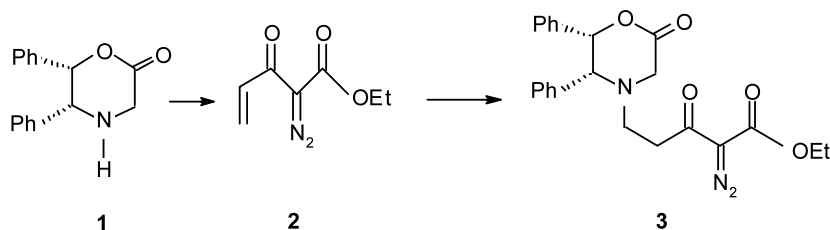
As regards chirality transfer, while in the case of the [2,3]-shift of spiro ammonium ylides important examples are known,² only one case of the [1,2]-Stevens rearrangement was reported by West and Naidu to proceed with remarkable diastereo- but modest enantioselectivity, in the key step synthesis of epilupinine alkaloids.³

Recently, following our interest in asymmetric synthesis through the carbenoid intramolecular reactivity,⁴ according to this cascade methodology, we prepared 5,7-condensed bicyclic 1,4-oxazepinones, with moderate diastereo-, but complete enantioselectivity, starting from 3-oxo-2-diazopentanoate *N*-linked to (3*R*)-6-oxo-3-phenyl-4-morpholine.⁵ The retention of configuration at the benzylic carbon atom involved in the [1,2]-ylide shift was observed.⁶

Herein is reported the preparation, by this protocol, of enantiopure nitrogen heterocycles of therapeutical interest.

To accomplish the [1,2]-sigmatropic rearrangement, we chose the diazocompound **3** in which the moiety bearing a diazo function is tethered to (2*R*,3*S*)-diphenylmorpholinone **1**. The latter has been previously employed, as a chiral template, in the Williams aminoacid syntheses (Scheme 1).⁷

The catalytic decomposition of **3** was expected to lead to the 5,7-bicyclic system **6** (Scheme 2), with better diastereoselectivity as in the case previously described.⁵ This expectation was supported, in addition of Williams' work, also by the better asymmetric induction obtained with **1**,⁸ if compared with the other chiral auxiliary mentioned.⁹ The asymmetric 3-prolinone and pyrrolidine targets will be respectively obtained through the lactonic ring opening of **6** by LiAlH_4 reduction and by a transesterification reaction, respectively.



Scheme 1.

Keywords: ammonium ylide; carbenoid; ring expansion; rearrangement.

* Fax: +39-79-229559; e-mail: saba@ssmain.uniss.it

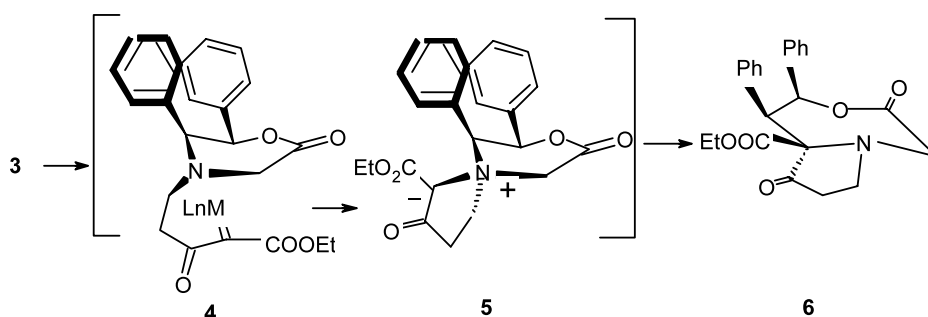
Both enantiomers and racemate of morpholinone **1** are commercially available as *N*-Boc- or *N*-Cbz- derivatives.¹⁰ Moreover, the pure enantiomer was obtained in a less expensive way according to the procedure reported by Brussee and co-workers, who started from the convenient chiral mandelonitrile.¹¹ The synthesis of ethyl 2-diazo-5-[(5*S*,6*R*)-5,6-diphenyl-2-oxomorpholin-4-yl]-3-oxopentanoate **3** was performed in (79%) yield, following the procedure previously described,⁴ by conjugate addition of **1** to ethyl 2-diazo-3-ketopent-4-enoate, **2**¹² (Scheme 1).

The decomposition of **3**, accomplished in boiling toluene, in the presence of 1% copper(II)acetylacetonate, was allowed to proceed until the diazo stretching band disappeared in the IR spectrum (15'); it quantitatively provided the rearranged ethyl-(1*S*,2*R*,9*aR*)-1,2-diphenyl-4,9-dioxohexahydropyrrolo[1,2-*d*][1,4]oxazepine-9*a*-(7*H*)-carboxylate, **6**, as a single diastereomer (Scheme 2).

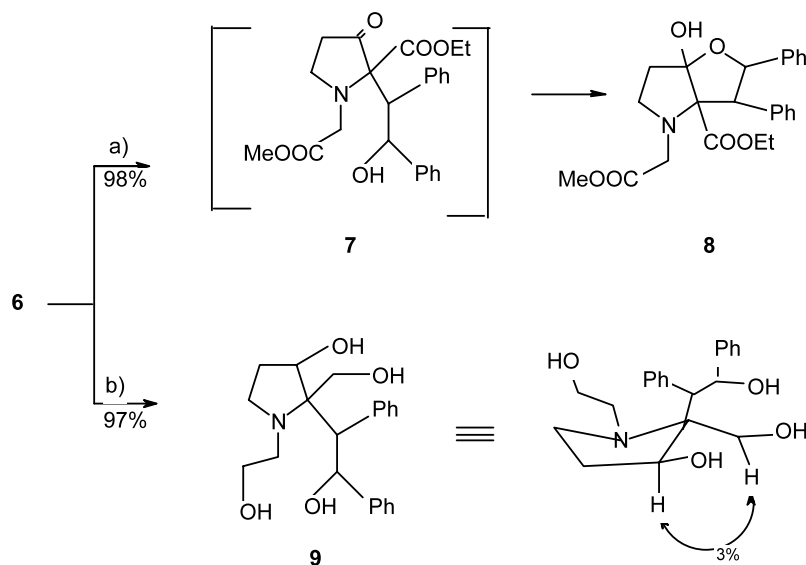
The superior catalytic activity of copper(II)-acetylacetonate for carbenic generation of onium ylides¹³ is confirmed. However, decomposition of **3**, performed with rhodium(II)-based catalysts, afforded complex reaction mixtures containing **6** as a minor

component. Transesterification of the lactone **6** in methanol, in the presence of TsOH, gave directly the bicyclic hemiacetal **8**, without traces of the expected 3-prolinone (Scheme 3). As a consequence of our failure to obtain crystals of compound **6** for an X-ray diffraction analysis, this was performed on compound **8** (Fig. 1).

Single-crystal X-ray analysis of the latter revealed the presence of a single enantiomer in the crystal, as shown by the space chiral group $P2_1$.¹⁴ This evidence confirmed the absolute structure of the 5,7-bicycle **6**. Particularly, the absolute configuration of the quaternary carbon bearing the ester group in *cis* position to the phenyl group at the neighbouring benzylic carbon has been established. Since the quaternary carbon is not affected in the ring opening process, its absolute configuration in compound **6** is identical to that in hemiacetal **8**. Moreover, LiAlH_4 reduction of **6** gave the substituted pyrrolidine **9**, as a single diastereomer in which the configuration of the quaternary carbon stereochemistry is established. Finally, the remaining absolute configuration assignment to the C-3 carbon of **9** was deduced from the 2D NOESY correlation, as depicted in Scheme 3. Consequently, the formation of



Scheme 2.



Scheme 3. Reagents and conditions: (a) TsOH, MeOH, rt, 3 h; (b) LiAlH_4 , THF, rt, 12 h.

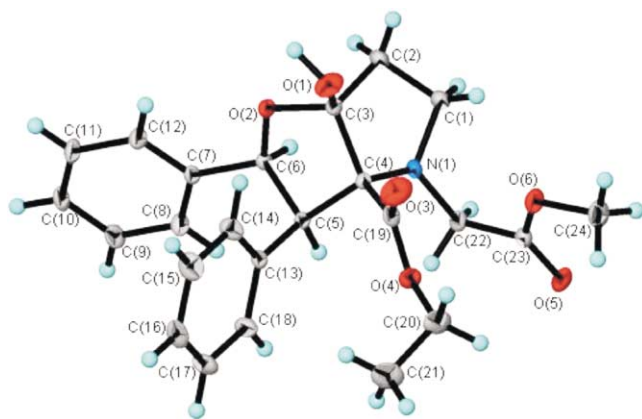


Figure 1. Atomic labelling scheme for compound **8**. The hydrogen atoms are labelled according to their parent atoms to which they are bonded (the ellipsoidal displacements are drawn at 40% probability level).

the intermediate spiranic ylide **5** can result from a carbenoid attack to nitrogen on the same face of both phenyl substituents. As a consequence of the complete retention of configuration at the migrating group, this process could still be considered intramolecular. However, a dissociation–recombination pathway involving a caged biradical could not be excluded, in spite of the fact that traces of products possibly arising from radical reactions were not detected, as in earlier examples of spiro ylides.¹⁵

In summary, herein is described a rapid and convenient construction of the bicyclic skeleton **6** in enantiopure form, through a cascade high asymmetric process with a radical key step.

The bicyclic 1,4-oxazepinone can be considered as a starting material to more complex systems like optically active alkaloids. Due to its convenient availability, this procedure can be extended to the (+)-enantiomer of the morpholinone **1**. Moreover, it is remarkable that in compounds **6**, **8** and **9**¹⁶ a quaternary carbon atom is present: the construction of molecules bearing asymmetric quaternary stereocentres has represented a very challenging and dynamic area in organic synthesis over the past decade.¹⁷

Finally, it is noteworthy that the enantiopure heterocycle **9** fits in the class of hydroxylated pyrrolidines, potential specific and competitive inhibitors of glycosidase enzymes with application in a range of medicinal areas including anti-HIV activity.¹⁸ The inhibition potency of this compound and its enantiomer against a variety of glycosidases will be reported in due course.

Acknowledgements

This work was supported by MIUR and Regione Autonoma della Sardegna. Thanks are due to Mr. Mauro Mucedda for experimental assistance.

References

- (a) Doyle, M. P.; McKervey, M. A.; Ye, T. *Modern Catalytic Methods for Organic Synthesis with Diazo-Compounds*; John Wiley & Sons: New York, 1997; (b) West, F. G.; Naidu, B. N. *J. Am. Chem. Soc.* **1993**, *116*, 1177; (c) Padwa, A.; Hornbuckle, S. F. *Chem. Rev.* **1991**, *91*, 263; (d) Padwa, A.; Beall, L. S. *Advances in nitrogen heterocycles*; JAI Press: Stamford, CT, 1998; Vol. 3, pp. 117–158; (e) Padwa, A.; Beall, L. S.; Eidell, C. K.; Worsencroft, K. J. *J. Org. Chem.* **2001**, *66*, 2414.
- (a) Clark, J. S.; Hodgson, P. B. *Tetrahedron Lett.* **1995**, *36*, 2519; (b) Wright, D. L.; Weekly, R. M.; Groff, R.; McMills, M. C. *Tetrahedron Lett.* **1996**, *37*, 2165; (c) Chappie, T. A.; Weekly, R. M.; McMills, M. C. *Tetrahedron Lett.* **1996**, *37*, 6523.
- (a) West, F. G.; Naidu, B. N. *J. Am. Chem. Soc.* **1994**, *116*, 8420; (b) Naidu, B. N.; West, F. G. *Tetrahedron* **1997**, *53*, 16565.
- (a) Chelucci, G.; Saba, A. *Angew. Chem., Int. Ed. Engl.* **1995**, *34*, 78; (b) Chelucci, G.; Saba, A. *Tetrahedron Lett.* **1995**, *36*, 4673; (c) Chelucci, G.; Saba, A.; Valle, G. *Tetrahedron: Asymmetry* **1995**, *6*, 807; (d) Chelucci, G.; Saba, A. *Tetrahedron: Asymmetry* **1997**, *8*, 699; (e) Chelucci, G.; Culeddu, N.; Saba, A.; Valenti, R. *Tetrahedron Lett.* **1999**, *40*, 8269.
- Chelucci, G.; Saba, A.; Valenti, R.; Bacchi, A. *Tetrahedron: Asymmetry* **2000**, *11*, 3449.
- For a general review of the Stevens rearrangement, see: Markò, I. E. In *Comprehensive Organic Synthesis*; Trost, B. M.; Fleming, I., Eds.; Pergamon: Oxford, 1991; Vol. 3, pp. 913–973.
- Williams, R. M. *Aldrichim. Acta* **1992**, *25*, 11.
- Vigneron, J. P.; Kagan, H.; Horeau, A. *Tetrahedron Lett.* **1968**, *9*, 5681.
- Tamura, M.; Harada, K. *Bull. Chem. Soc. Jpn.* **1980**, *53*, 561.
- Commercially available from Aldrich Chemical Co.: (2*R*,3*S*)-(–)-*tert*-butyl-6-oxo-2,3-diphenyl-4-morpholine carboxylate (cat. no. 33-184-8).
- van den Nieuwedijk, A. M. C. H.; Warmerdam, E. G. J. C.; Brussee, J.; van der Gen, A. *Tetrahedron: Asymmetry* **1995**, *6*, 801.
- Zibuck, R.; Streiber, J. M. *J. Org. Chem.* **1989**, *54*, 4717.
- (a) West, F. G.; Naidu, B. N.; Tester, R. W. *J. Org. Chem.* **1994**, *59*, 6892; (b) Clark, J. S.; Krowiak, S. A.; Street, L. J. *Tetrahedron Lett.* **1993**, *34*, 4385.
- Crystal data and structure refinement for **8**: single crystals were obtained by recrystallization from ethyl acetate. The substance (C₂₄H₂₇NO₆, *M_r*=425.47) crystallized in the monoclinic space group *P*2₁, *a*=7.6394(15), *b*=12.909(3), *c*=11.262(2) Å; α=90°, β=108.57°, γ=90°; *V*=1052.8(4) Å³, *Z*=2, *D*_{calcd}=1.342 Mg/m³, *F*(000)=452, *T*=173 K, crystal size 0.40×0.30×0.18 mm, θ=3.23° to 28.30°, limiting indices = –10 ≤ *h* ≤ 10, –17 ≤ *k* ≤ 16, –15 ≤ *l* ≤ 14, reflections collected=4888, unique=4888 (*R*_{int}=0.000), completeness to θ=28.30:99.6, refinement method: full-matrix least-squares on *F*², data=4888, restraints=2, parameters=286, goodness-of-fit on *F*²=1213, final *R* indices [*I* > 2σ(*I*)] *R*₁=0.0561, *wR*₂=0.1327, *R* indices (all data): *R*₁=0.0591, *wR*₂=0.1365, absolute structure parameter=–0.3(8), extinction coefficient=0.379(19), largest diff. peak and hole=0.854 and –0.882 e Å^{–3}.

Crystallographic data have been deposited with the Cambridge Crystallographic Data Centre (Deposition Number: CCDC 195180).

15. Ollis, W. D.; Rey, M.; Sutherland, I. O. *J. Chem. Soc., Perkin Trans. 1* **1983**, 1009.
16. *Data for selected compounds*: Compound **3**: Pale yellow crystals, mp 126–127°C; $[\alpha]_D^{25} = +66$ (*c* 0.9, CHCl₃); ¹H NMR (300 MHz, CDCl₃): δ 7.18–7.09 (m, 6H), 7.04–7.01 (m, 2H), 6.88–6.85 (m, 2H), 5.71 (d, 1H, *J*=4.2 Hz), 4.27 (q, 2H, *J*=7.2 Hz), 4.21 (d, 1H, *J*=4.2 Hz), 3.76 (AB system, 2H), 3.14–2.96 (m, 2H), 2.90–2.73 (m, 2H), 1.31 (t, 3H, *J*=7.2 Hz); ¹³C NMR (CDCl₃): δ 190.9, 168.7, 161.1, 135.4, 134.1, 129.4, 127.9, 127.8, 126.7, 82.9, 65.5, 61.5, 51.9, 49.4, 37.1, 14.3; IR (Nujol): 2950, 2920, 2850, 2140, 1740, 1710, 1660, 1460, 1372, 1295, 1210, 1130, 1050, 715 cm⁻¹. Anal. calcd for C₂₃H₂₃N₃O₅: C, 65.55; H, 5.50; N, 9.97. Found: C, 65.21; H, 5.71; N, 9.65.
- Compound **6**: White plates, mp 64–65°C; $[\alpha]_D^{25} = +77$ (*c* 1.0, CHCl₃); ¹H NMR (300 MHz, CDCl₃): δ 7.30–7.06 (m, 10H), 5.87 (s, 1H), 4.39 (q, 2H, *J*=7.8 Hz), 4.02 (AB system, 2H), 3.62 (dd, 2H, *J*=9.0, 3.6 Hz), 2.41 (t, 1H, *J*=3.9 Hz), 2.35 (t, 1H, *J*=3.9 Hz), 2.17–2.09 (m, 2H), 1.39 (t, 3H, *J*=7.8 Hz); ¹³C NMR (CDCl₃): δ 207.4, 169.9, 169.5, 138.1, 132.6, 131.2, 127.8, 127.7, 127.6, 125.7, 79.6, 68.4, 62.5, 52.7, 52.3, 49.0, 37.0, 14.3; IR (Nujol): 3849, 3419, 2360, 1742, 1601, 1522, 1492, 1455, 1376, 1334, 1302, 1222, 1177, 969, 854, 779, 756 cm⁻¹. Anal. calcd for C₂₃H₂₃NO₅: C, 70.21; H, 5.89; N, 3.56. Found: C, 70.42; H, 6.01; N, 3.22.
- Compound **8**: White crystals, mp 137–138°C; $[\alpha]_D^{25} = -25$ (*c* 0.9, CHCl₃); ¹H NMR (300 MHz, CDCl₃): δ 7.28–6.94 (m, 10H), 5.92 (s, 1H), 5.675 (d, 1H, *J*=6.0 Hz), 3.83 (AB system, 2H), 3.78 (q, 2H, *J*=6.9 Hz), 3.74 (s, 3H), 3.57 (d, 1H, *J*=6.0 Hz), 3.19 (t, 2H, *J*=6.9 Hz), 2.42 (2H, AB system), 0.76 (t, 3H, *J*=6.9 Hz); ¹³C NMR (CDCl₃): δ 171.6, 170.9, 137.9, 136.2, 130.2, 127.5, 127.4, 126.5, 126.3, 126.0, 114.0, 81.89, 61.0, 57.2, 51.8, 50.6, 49.7, 39.8, 13.2; IR (Nujol): 3270, 1727, 1692, 1457, 1371, 1343, 1264, 1188, 1120, 1092, 1056, 988, 964, 848, 767, 718, 701 cm⁻¹. Anal. calcd for C₂₄H₂₇NO₆: C, 67.75; H, 6.40; N, 3.29. Found: C, 67.51; H, 6.28; N, 3.60.
- Compound **9**: White crystals, mp 189°C; $[\alpha]_D^{25} = -90$ (*c* 0.9, CH₃OH); ¹H NMR (300 MHz, CD₃OD): δ 7.20–6.95 (m, 10H), 5.67 (s, 1H), 4.86 (s, 4H), 4.40 (dd, 1H, *J*=5.7, 4.8 Hz), 3.67–3.63 (m, 2H), 3.47–3.43 (m, 2H), 3.21–3.19 (m, 2H), 3.01–2.92 (m, 1H), 2.75 (dt, 1H, *J*=9.0, 2.4 Hz), 2.68 (dt, 1H, *J*=5.4, 12.6 Hz), 2.34–2.23 (m, 1H), 2.0–1.88 (m, 1H); ¹³C NMR (CD₃OD): δ 145.8, 137.8, 133.7, 128.3, 128.2, 127.5, 127.4, 75.8, 72.3, 64.2, 61.3, 59.5, 52.1, 50.7, 33.1; IR (Nujol): 3816, 3707, 3668, 3646, 3563, 3495, 2362, 1646, 1600, 1492, 1374, 1340, 1288, 1261, 1229, 1204, 1174, 1142, 1100, 1086, 993, 963, 734, 611 cm⁻¹. Anal. calcd for C₂₁H₂₃NO₄: C, 71.37; H, 6.56; N, 3.96. Found: C, 71.62; H, 6.36; N, 4.12.
17. (a) Fujii, K. *Chem. Rev.* **1993**, 93, 2037; (b) Corey, E. J.; Guzman-Perez, A. *Angew. Chem., Int. Ed. Engl.* **1998**, 37, 388; (c) Dehli, J. R.; Gotor, V. *J. Org. Chem.* **2002**, 67, 1716.
18. (a) Elbein, A. D. *Ann. Rev. Biochem.* **1987**, 56, 497; (b) Legler, G. *Adv. Carbohydr. Chem. Biochem.* **1990**, 48, 319.